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An acid is a molecule or ion capable of donating a hydron (proton or hydrogen ion H⁺), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or Brønsted acids. In the special case of aqueous solutions, proton donors form the hydronium ion H₃O⁺ and are known as Arrhenius acids.

Acid - Wikipedia

References (Click the next to a value above to see complete citation information for that entry). Allred, A. L. "Electronegativity Values from Thermochemical Data." Journal of Inorganic and Nuclear Chemistry, volume 17, number 3-4, 1961, pp. 215–221. doi:10.1016/0022-1902(61)80142-5 Allred, A. L., and E. G. Rochow.

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Il punto fermo di questa teoria è che una sostanza si comporta come acido soltanto quando è in presenza di una sostanza capace di comportarsi come base nei suoi confronti e viceversa.

Teoria acido-base di Brønsted-Lowry - Wikipedia

Ominaisuuksia. Gammateily on kaikkein suurienergisiin sähkömagneettista teily. Yksittäisen gammakvantin energia on tyypillisesti yli 100 keV, taajuus luokkaa 10²⁰ Hz ja aallonpituus alle 10 pm. Pienen aallonpituutensa vuoksi gammakvantin keskimääräinen vapaa matka väliaineessa on suuri ja siksi se on hyvin läpäitunkevä. . Gammateilyn tehon vähentämiseksi 50 %:iin ...

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